Insight into the hydrothermal stability of mesostructured cellular silica foams

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The mesostructured cellular silica foams (MCFs) is a kind of promising catalytic materials especially for bulky molecular reaction [1-2]. However, the hydrothermal stability of mesoporous materials is an important factor limiting the extensive industrialized application [3-4]. Here, the hydrothermal stability of MCFs was investigated detailedly for the first time. It was characterized using transmission electron microscopy (TEM), nitrogen sorption, ²⁹Si solid-state nuclear magnetic resonance (²⁹Si NMR) and Fourier transform infrared (FT-IR). We found more micropores and "round-like" pores contribute to the stability of MCFs prepared at 550 °C in high temperature steam. During the initial 3 h steam treatment in 600 °C, many specific surface area and pore volume were lost. Simultaneously, the polymerization degree of ≡Si-O-Si≡ linkage was largely enhanced and the framework of MCFs prepared at 550 °C became more "stout". After this treatment, the change of the structural parameter of MCFs was slight with the elongation of the steamed time. The framework of MCFs prepared at 550 °C can not withstand the more critical condition of 800 ^oC steam and collapsed completely. By elevating the calcination temperature of MCFs to 900 ^oC, the polymerization degree of framework was further enhanced. Hence, it exhibited high hydrothermal stability under the steam of 800 °C. When the steam treatment was prolonged, the specific surface area and pore volume decrease gradually, but the mesostructure of MCFs was well-retained. Furthermore, we found that the variation characteristics between MCFs and SBA-15 [3] were distinct under the high temperature steam. We make a qualitative illustration on the discriminating phenomena from the thermodynamic standpoint.

[1]. M. Cai, J. Sha, Q. Xu, Tetrahedron 63, 4642 (2007).

[2]. A. Indra, S. Basu, D. G. Kulkarni, C. S. Gopinath, S. Bhaduri, G. K. Lahiri, Applied Catalysis A: General **344**, 124 (2008).

[3]. F. Zhang, Y. Yan, H. Yang, Y. Meng, C. Yu, B. Tu, D. Zhao, J. Phys. Chem. B **109**, 8723 (2005).

[4]. D. T. On, S. Kaliaguine, J. Am. Chem. Soc. 125, 618 (2003).